



COMPARATIVE STUDY OF ARSENIC AND CADMIUM COMBINED METAL TOXICITY IN FRESH WATER FISH WITH YAMUNA WATER FISH

ARTI SRIVASTAVA, SOHINI SINGH* AND TANUALLEN

Amity Institute of Biotechnology, Amity University, Noida, India

ABSTRACT

Yamuna River is one of the most important natural water resources essential for survival and growth. Over exploitation of this resource has made it one of the most polluted rivers in the world. Heavy metals such as Arsenic and Cadmium enter the aquatic environment due to rapid industrialization and other anthropogenic activities. High cadmium level definitely harms the fish health, since the arsenic present also tends to bioaccumulate in the fish organs, there are possibilities of alteration in toxicity due to the presence of both these metals together. Hence the objective of present study was designed to compare the toxicity due to two metals present together in Yamuna water on the fish *Channa punctatus*. Accumulation of metals was significantly higher ($p < 0.05$) in metals in combination as compared to Yamuna water fish with control while lipid peroxidation was significantly higher in Yamuna water fish. Level of catalase, SOD, GSH and GST were significantly lower in both Yamuna water fish and metal treated fish compared to control.

KEYWORDS: Arsenic, Cadmium, Yamuna river, Toxicity



SOHINI SINGH

Amity Institute of Biotechnology, Amity University, Noida, India

INTRODUCTION

Rapid industrialization, growth, increased demand for resources and materials have led to the exploitation of natural resources^{1, 2}. Yamuna one of the most important water body for survival and growth originating from Yamunotri, the main tributary of the river Ganga is turning into one of the most polluted rivers in the world, especially the Delhi stretch where city dumps about ~58% of its waste into it³. Elevated level of various heavy metals was reported in the river and land around Delhi regions^{4, 5}. Heavy metals such as arsenic (As) and cadmium (Cd) enters the aquatic environment by mining, burning of fossil fuels, pesticide application, industrial activities like electroplating, fabrics, plastics, ceramic and glass, vehicle and electronics^{6,7,8} which can be toxic to the aquatic organism in even a very small amount. These metals when present in water bodies were usually get bioconcentrated, bioamplified and bioaccumulate in the tissue of aquatic organisms through which they become the part of various food chains and prove to be toxic to both humans and other organisms^{9,10}. Oxidative stress is the major consequence of water pollution in aquatic organisms. Fish are endowed with an antioxidant defense system to protect them from this stress¹¹. However when there is imbalance in antioxidant defences, leading to the generation of reactive oxygen species (ROS), which can be detoxified by metallothioneins (MT), superoxide dismutase (SOD), catalase (CAT) and glutathione related enzymes. Antioxidant enzymes level can be used as a biomarker of oxidative stress subsequent to pollutants exposure in aquatic animals^{12, 13}. Arsenic and cadmium result in induction of antioxidant defense mechanisms^{14,15}. Being a major source of food and a part of the very high healthy diet for human; fish have great demand in market. They were, however, going to be in danger mainly because of water pollutants¹⁶. Fish can be used as bioindicator of any aquatic health system as they occupy the top of an aquatic food chain and constitute an excellent model to understand mechanistic aspects of metal toxicity¹⁷. According to ATSDR (Agency for

Toxic Substances and Disease Registry), of the U.S. Department of Health and Human Services arsenic and cadmium are listed as the first and seventh most commonly found toxic substance in the United States. Due to their known or suspected toxicity these metals are established to pose the most significant potential threat to human health. Central Water Commission (CWC)¹⁸ have reported the status of toxic metal content of Indian Rivers and found that arsenic concentration was within the acceptable limit while cadmium content was more than the acceptable limits of Bureau of Indian Standard (BIS) in river Yamuna at Delhi Rly Bridge. Cadmium was found above the acceptable level of Bureau of Indian Standard in Yamuna water obtained from Delhi-Nizammudin Bridge¹⁹. High cadmium level definitely harms the fish health, since arsenic is also there to bioaccumulate in the fish organs, there are possibilities of alteration in toxicity due to the presence of both these metals together. Hence the present study was designed to compare the toxicity due to two metals in combination with Yamuna water.

MATERIALS AND METHODS

Healthy specimens of fresh water Murrel, *Channa punctatus* (Bloch) were procured from fish farm. They were acclimatized individually in all glass aquaria in dechlorinated water at laboratory conditions ($25 \pm 2^\circ\text{C}$) for 1 month. During acclimatization fish were fed with goat liver. For Yamuna water analysis surface water samples were collected from Okhla barrage and taken to the laboratory for the estimation of pH, temperature, and dissolved oxygen (DO) and biochemical oxygen demand (BOD). Reagent bottles of 250ml were used for collection of water samples. For dissolved oxygen (DO), we used 1 ml of each Winkler's solutions A (Manganous sulphate) and B (Potassium hydroxide in Potassium iodide) to fix the oxygen content of the water. For BOD, water samples were immediately wrapped in black polyethylene bags and transported to the laboratory. Fish were caught with the help of local fisherman, preserved in ice and transported to the laboratory. In laboratory liver

tissues and scales were collected and washed with deionized water then preserved at -80°C for further analysis. Average wet weight (g) and length (cm) of both Yamuna water fish and laboratory maintained fish were 19.5 ± 2.25 gm and 11.5 ± 0.2 cm respectively.

Test chemicals

For the present study all reagents and chemicals of highest purity were used. Analytical-grade arsenic tri-oxide (As_2O_3) (99.5%) and cadmium chloride ($\text{CdCl}_2 \cdot \text{H}_2\text{O}$) (99%), phenazine methosulphate and nitro-blue tetrazolium manufactured by Central Drug House (P) Ltd. have been used. Thiobarbituric acid was purchased from Loba Chemie (Mumbai). 5-5-Dithiobis-2-nitrobenzoic acid, 1-Chloro-2, 4-dinitrobenzene, glutathione, bovine serum albumin and tricaine methanesulfonate were purchased from Sigma Chemical Co. (St. Louis, MO, USA). Hydrogen peroxide was purchased from Fisher Scientific. Tetra- Sodium pyrophosphate was purchased from Merck Specialities (P) Ltd Mumbai.

Experimental Procedure

The physicochemical properties of test water, viz. temperature, pH, dissolved oxygen, chloride, were analyzed using standard methods. Before starting the experiment the median lethal concentration of heavy metals for 96 h (96 h LC₅₀ value) was calculated by pursuing the 24 h renewal bioassay system and Trimmed Spearman-Kärber method²⁰. The acclimatized fish were divided into 4 groups. Group A fish were introduced to an aquarium containing Arsenic tri oxide (1 mg/ml, 10% of 96 h LC₅₀). Group B fish were introduced with Cadmium chloride (0.8 mg/L, 10% of 96 h LC₅₀) and group C fish were introduced to a combination of Arsenic tri-oxide + Cadmium chloride (1 mg/L + 0.8 mg/L) and group D fish were treated as a control (without any metal). Concentration of metals in the water of experimental aquaria was kept constant. A record of their mortality was also maintained. All groups contained five fishes with 20 L of water in each aquarium. Experiment was performed in triplicates.

Sample preparation

After exposure of 15 days, the fish from each group were euthanized; liver tissues and scales were collected. Samples were preserved at -80°C for further analysis.

1. Scales

Scales were removed from Yamuna water fish (mentioned above), heavy metal treated (individual and combination groups) and control fish. They were stained with Borax Carmine and chromatophore patterns were observed under the light microscope²¹.

2. Lipid Peroxidation assay

The lipid peroxidation in the liver tissue was assayed by determining microsomal malondialdehyde²².

3. Protein concentration

Protein was estimated in liver tissue by method of Lowry²³ using bovine serum albumin as the standard.

4. Estimation of Metals

To analyze the metal concentration in liver tissue, tissue was digested in concentrated nitric acid and then diluted with double distilled water. 50 mg of liver tissue was digested in 1 ml of concentrated nitric acid at 80°C for 1 h and it was evaluated through Atomic Absorption Spectrophotometer (Perkin Elmer AA800).

5. SOD activity

SOD activity was analyzed using the method described by²⁴. Reaction mixture contained 2 ml sodium pyrophosphate buffer (pH8.3), 0.1 ml phenazine methosulphate, 0.2 ml nitro blue tetrazolium and 0.2ml tissue supernatant. Reaction was initiated by adding 0.2 ml NADH at 30°C and stopped by addition of 2 ml glacial acetic acid. Color intensity of the solution was measured at 560 nm. The SOD activity was expressed as % of inhibition.

6. Catalase activity

Catalase activity was quantified with the procedure described by²⁵. The reaction

mixture contained 1.96 ml of 50 mM sodium phosphate buffer (pH 7.0) and 1 ml of 40 mM H₂O₂. The reaction was initiated by adding 40 µl of tissue supernatant. The CAT activity was determined by measuring the rate of disappearance of H₂O₂ at 340 nm for 1 min. Enzyme activity was expressed as I M H₂O₂ decomposed/min/mg protein.

in ethanol and the enzyme solution. GST activity was expressed as nM cDNB conjugates/min/mg of protein

Statistical analysis

Statistical analysis was done using SPSS.

RESULTS

The physiochemical properties of the experimental water and Yamuna water were measured in which Dissolved Oxygen was 7.34 mg/l for experimental water and 1.61 mg/l for Yamuna water. The temperature of experimental water and Yamuna water was same 18.0 - 20.0 °C. The pH values for experimental water and Yamuna water were 7.0 – 7.1 and 7.8 – 8.0 respectively. As shown in Fig 1 accumulation of metals (arsenic and cadmium) in liver of Yamuna water fish was significantly higher (p<0.05) than the liver tissue of control fish whereas it was lower when compared to the individual metal (arsenic and cadmium) treated group. Accumulation of arsenic was found to be more when given individually than compared to combination. Same observations were also found in group of cadmium treated fish.

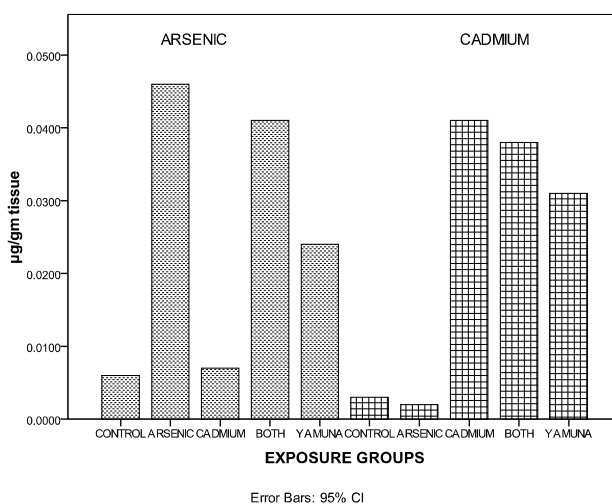
7. Reduced Glutathione (GSH)

Amount of reduced glutathione in the liver was estimated by using Ellmans' reagent²⁶. Reaction of GSH and DTNB (5-5-dithiobis-2-nitrobenzoic acid) was resulted in the formation of a yellow colored product 2-nitro-5-thiobenzoic acid. GSH concentration was determined by measuring the absorbance at 412 nm. The unit of glutathione was expressed as nM GSH /mg protein.

8. GST activity

GST activity was calculated spectrophotometrically at 30°C by following the formation of GSH conjugate with 1-chloro-2,4-dinitrobenzene (cDNB) at 340 nm using an extinction coefficient of 9.6 mM⁻¹ cm⁻¹²⁷. The reaction mixture contained 2 ml volume of 0.1 M sodium phosphate buffer (pH 6.5), 50 mM GSH, 50 mM cDNB

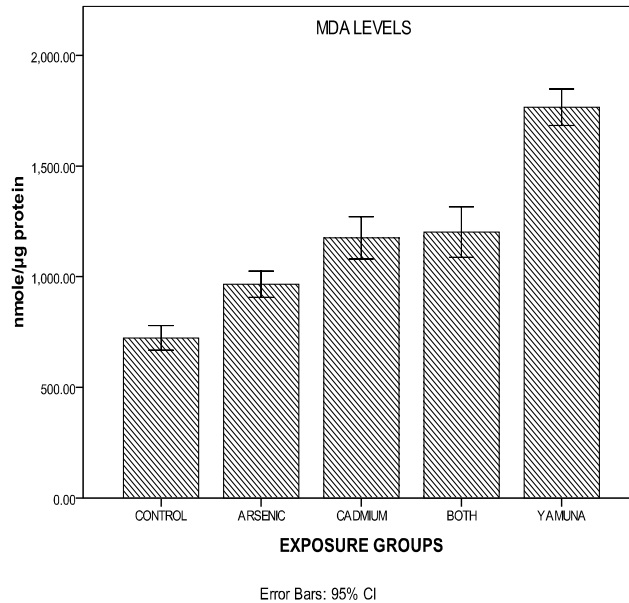
Figure 1
Metal concentration in liver tissue



Exposure of arsenic and cadmium individually for 15 days significantly ($p < 0.05$) increased the level of lipid peroxidation in the liver as compared to the control group. Same observation was also found with Yamuna water

fish (Fig 2). Lipid peroxidation was significantly higher in Yamuna water group when compared with individual and combined metals group and control group.

Figure 2
MDA levels after exposure. $n = 5$ in each exposure; values are represented as mean \pm SE, $p < 0.05$



The antioxidant enzymes like SOD and CAT decreased significantly ($p < 0.05$) in metals treated group and Yamuna water group.

Moreover, decrease in SOD and CAT level in Yamuna water fish is more significant than metal treated group.

Figure 3
SOD activity after exposure. $n = 5$ in each exposure; values are represented as mean \pm SE, $p < 0.05$

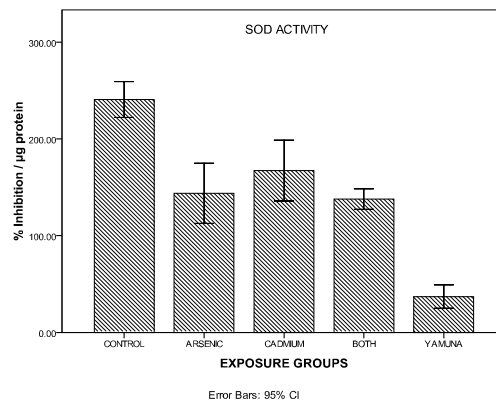
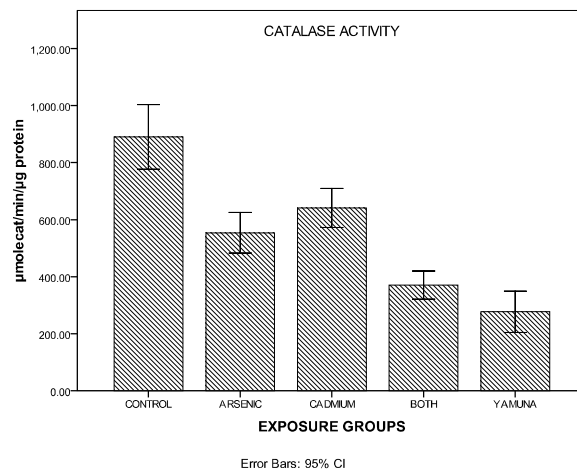


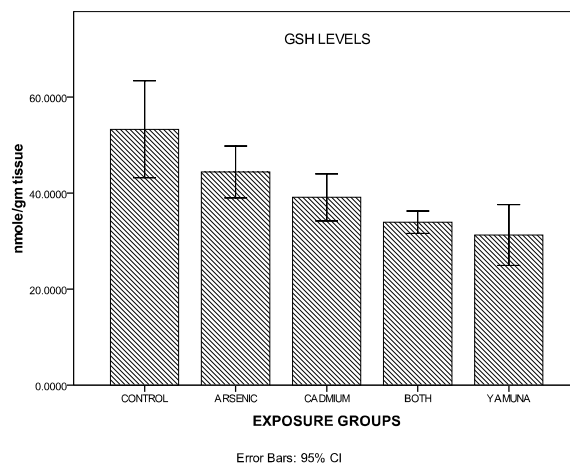
Figure 4
Catalase activity after exposure. n = 5 in each exposure;
values are represented as mean ± SE, p < 0.05



The result shown in the Figure 5 shows significantly reduced level of GSH ($p < 0.05$) in fish exposed to metal arsenic, cadmium individually, in combination and in Yamuna

water fish as compared to control. However, decline in reduced glutathione is more significant in Yamuna water fish than metal treated group individually.

Figure 5
GSH levels after exposure. n = 5 in each exposure;
values are represented as mean ± SE, p < 0.05



Significant ($p < 0.05$) reduced level of GST enzyme were observed in fresh water fish *Channa punctatus* exposed to As, Cd

individually, in combination and in Yamuna water fish when compared with control.

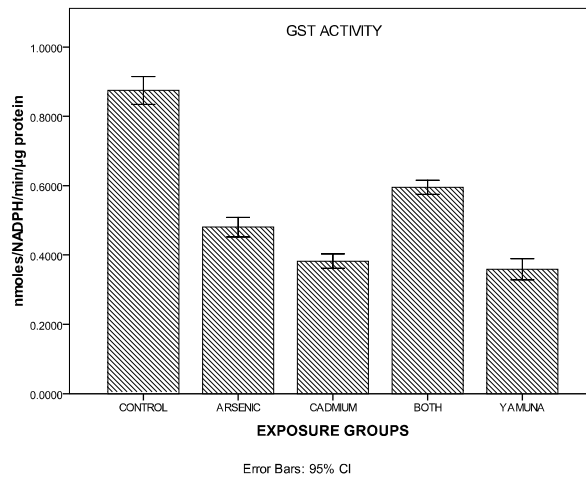


Figure 4
GST activity after exposure. n = 5 in each exposure; values are represented as mean ± SE, p < 0.05

Accumulation of metals in the scales produces the variation in the chromatophores position, shape and density which was shown in the respective figures. In the control group the chromatophores were punctuated and densely present. However, in arsenic treated group number of chromatophores was reduced, but reticulated and a specific linear pattern was

visualized. In cadmium treated groups, the chromatophores were very less in number and dispersed throughout the scale. In combined metals group, the number of chromatophores reduced. In Yamuna water fish, chromatophores had maximum variation in shape, size and distribution.

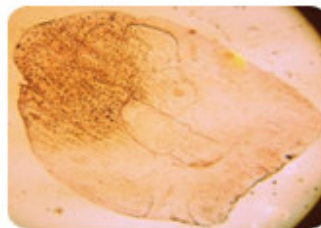


Figure 8
Scale of control fish



Figure 9
Scale of cadmium treated fish

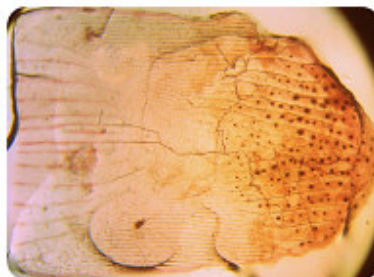


Figure 10
Scale of Arsenic treated group

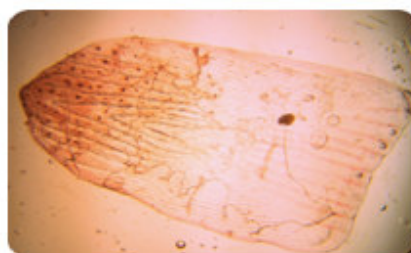


Figure 11
Scales treated with both Arsenic and Cadmium

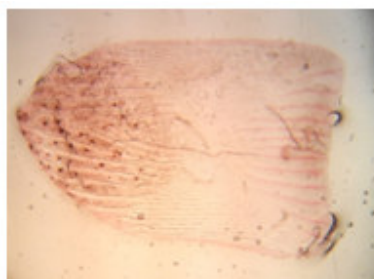


Figure 12
Scales treated with Yamuna water

DISCUSSION

The present work was intended to study the level of heavy metals and certain biomarkers of oxidative stress in fish caused by exposure of metals in laboratory condition, and in Yamuna water fish *Channa punctatus*. Our results clearly indicated the accumulation of metals in *Channa punctatus* which caused an elevation of malondialdehyde and alternation in the other antioxidant enzymes in fish. Exposures to certain metals are effective inducer of oxidative stress, furthermore, they have toxic and carcinogenic effects^{28, 29}. The liver is the major

target organ of inorganic arsenic, which is explained by the affinity of As (III) toward vicinal dithiol in hepatic cytosolic proteins. The binding of arsenic in target organs was postulated as a first step of arsenic detoxification, this binding is also responsible for the major intoxication symptoms: hepatic and renal failure and cardiovascular and neurological effects³⁰. Birben³¹ enlightened oxidative stress, one of the major consequences of elevated levels of metal toxicity, is a well-known mechanism for the production of reactive oxygen species (ROS). It occurs when there is an imbalance between productions of ROS and the cellular antioxidant

defense system which results in formation of superoxide anion radicals, hydrogen peroxide, hydroxyl radical, alkoxy radical, peroxy radical. Oxidation of proteins and lipids, modification in gene expression, and changes in cell redox status is aftermath of metal exposure through ROS mediated reactions^{32,17}. Malonyldialdehyde is a complex secondary end-product of lipid degradation induced by metals, perpetuate the disruption of cell integrity hence, its presence in tissue showed a reliable biomarker of oxidative stress^{33, 34}. Therefore we included the evaluation of MDA content in the liver of metal treated and Yamuna water fish. The results of present study clearly revealed that an exposure of arsenic and cadmium individually for 15 days, significantly ($p < 0.05$) increased the level of lipid peroxidation in the liver as compared to the control group. Same observation was also found with Yamuna water fish. Elevation in the lipid peroxidation was significantly higher in Yamuna water group than the metal treated group and control. Accumulation of metals (arsenic and cadmium) in Yamuna water fish was although significantly higher than the control, it was lower when compared to the arsenic and cadmium treated group. The increase in lipid peroxidation in Yamuna water fish may be attributed to the presence of other pollutants in the Yamuna water. Our result was supported by a previous study performed on different fish tissue in Yamuna water^{35, 36}. Increased level of LPO was observed by many investigators in heavily polluted sites. Ferreira and Bacanskas^{37, 38} found elevated level of lipid peroxidation from polluted site of river Douro Estuary, Portugal in two resident species, mullet (*Mugil cephalus*) and flounder (*Platichthys flesus*) and in killifish (*Fundulus heteroclitus*) inhabiting a polluted inlet of the Elizabeth River respectively. Induction of TBARS content in three-spined stickleback (*Gasterosteus aculeatus* L.) have been observed, impacted by various sources of contaminations in comparison to fish from uncontaminated sites³⁹. Superoxide-dismutase (SOD) is antioxidant metalloenzyme which reduces superoxide radicals to water and molecular oxygen⁴⁰ and consequently catalase

(CAT) reduces hydrogen peroxide to molecular oxygen and water⁴¹. Since both enzymes are linked functionally therefore, similar parallel trend in the activities of both enzymes should be there. In support to this, we found that at an exposure of 15 days, SOD and CAT activities decreased significantly ($p < 0.05$) in metals treated group and Yamuna water group as compared to control, although, decrease in SOD and CAT level in Yamuna water fish is more significant than metal treated group indicating more damage to the antioxidant enzymatic defence in Yamuna water fish. Similar findings were also observed by some other investigators^{42, 43}. They found decreased SOD and CAT activity impacted with toxicity of other metals. Farombi⁴⁴ also observed a decrease in CAT activity in *Clarias gariepinus* from Nigeria Ogun River. In contrast to our result^{45,46} have reported hepatic SOD activity decreased with increased level of CAT which might be because decreased SOD level demonstrated damage in antioxidant mechanism due to metal exposure. In contrary to this, increase in CAT showed its participation to cope up with elevated level of oxidative stress as shown by⁴⁷. Reduced glutathione has been considered as nonenzymatic antioxidant, pertaining important cellular defence against oxidative stress and products formed due to oxidative damage⁴⁸. Glutathione sulfotransferase (GST), an imperative glutathione associated enzyme catalyzes the conjugation of GSH with various electrophilic substances to detoxify various reactive intermediates and keep equilibrium between ROS production and elimination^{45, 49}. There were disparity in the results observed by various workers in the content of GSH and GST in fish tissue exposed to metals where it was found due to their organ specific response^{50, 51, 52, 53}. Results from present study showed significant reduced level of GSH and GST after 15 days, in fresh water fish *Channa punctatus* exposed to As, Cd individually, in combination and in Yamuna water fish as compared to control. Reduction in the GSH level can be explained by their conjugation with metals, hence declining the total GSH pool available⁵⁴. Results obtained indicated that GSH might

have participated in protection against oxidative stress due to heavy metal toxicity and consequently decreased sharply in its level. Atli and Canli, Lima^{50, 55} observed a significant increase in level of GSH at exposure of Cd, Cu and contaminated effluents respectively in liver of *O. niloticus*, whereas Cao⁵⁶ investigated decreased level of reduced glutathione and GST in Japanese flounder juveniles due to cadmium exposure in liver, while Jia⁵⁷ found increased level of GSH in liver of Oujiang colored common carp *Cyprinus carpio* Var. color. Present study clearly gives a view of oxidative stress in Yamuna water fish when compared with other treatment groups. The metals hampered the antioxidant defense hence reduced antioxidants activity was observed in fish of metal treated individually and combined group. Compared to metal treated groups, Yamuna water fish had higher level of lipid peroxidation indicating more oxidative stress to the fish. While, their antioxidant defense is not reduced in the same manner. Thus definitely showing a better antioxidative defense in Yamuna water fish. Chromatophores are specialized cells which can produce pigments and are originated in neural crest during embryonic development. Depending on the species, they have various types of hormone receptors and neurotransmitters which participate in aggregation and dispersion of chromatophores^{58, 59}. Considering this fact fish chromatophores were also observed in this study to perceive change in pigmentation due

to altered surrounding conditions. In the present study, there was marked changes in fish chromatophores noticed in both Yamuna water and metal treated group. Accumulation of metals in the scales produced the variation in the chromatophores position, shape and density. Our study reports that in arsenic treated group, numbers of chromatophores were reduced but they become reticulated and there was also a specific linear pattern visualized. In cadmium treated group, the chromatophores were very less in number and dispersed throughout. In combined metals group, the chromatophores greatly reduced and almost disappeared showing the degeneration of chromatophores. Yamuna water group has shown maximum variation in shape, size and density. This variation is again attributed to variety of stress Yamuna water fish have to face. Similarly Bajpai and Tripathi, Kaur and Dua^{60,61} also found dispersion of chromatophores in fresh water fish *Channa punctatus* and an alteration in the shape, density and dimension after fluoride exposure in stinging catfish *Heteropneustes fossilis* (Bloch) respectively from water channel Tung Dhab drain. Due to metal toxicity chromatophores number reduced greatly. Thus it can be predicted in order to survive in the toxic environment, the fish might have maintained its antioxidants defense and hence chromatophores are not entirely damaged or reduced speculating their protection against toxicity to survive in pollutants loaded environment.

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