



ONE POT SYNTHESIS OF 3-(SUBSTITUTED PHENOXYMETHYL)-6-PHENYL/SUBSTITUTED PHENOXYMETHYL-1,2,4-TRIAZOLO[3,4-B][1,3,4]THIADIAZOLE DERIVATIVES AS ANTIMICROBIAL AGENTS

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ABSTRACT

The reaction of thiocarbohydrazide with substituted phenoxy acetic acid to obtained 3-substituted-4-amino-5-mercapto-1,2,4-triazoles (1) by condensing compound 1 with aromatic carboxylic acid and substituted phenoxy acetic acids resulted the title compound triazolothiadiazoles 2a-j. The purity of the newly synthesized compounds was confirmed by TLC. Structures of all the newly synthesized compounds were confirmed by spectral data. All the newly synthesized compounds were screened for their *in-vitro* antimicrobial activity. Among the series the compounds 2e, 2f and 2b, 2c, 2f, 2d, 2i were exhibited equipotent antibacterial and antifungal activity at MIC of 1 µg/mL when compared with standard drugs respectively. From the results the compounds 2c, 2f, 2j were showed comparable antitubercular activity against *M. tuberculosis* H₃₇R_v at MIC of 0.50 µg/mL, when compared with standard drug Rifampin and INH which showed MIC of 0.25 µg/mL.

KEYWORDS: 1,2,4-triazoles, triazolothiadiazoles, antimicrobial activity, antibacterial activity, antifungal activity, antitubercular activity



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INTRODUCTION

Microbial infections are the major threat in the world; these infections are treated with different antimicrobial agents. Resistance to antimicrobial agents by microorganism was recognized from the beginning¹. The spread of this phenomenon is abetted by short generation time and genetic versatility of microorganisms, as well as by poor antibiotics prescribing and utilization practices. The antibiotics prescribed have the different adverse effects, such as; sulphonamides, fluoroquinolones and β -lactam antibiotics are allergic to some individuals, cephalosporin and polyene antibiotics share common nephrotoxicity², aminoglycoside antibiotics exert ototoxic, antitubercular drugs cause progressive liver damage and antiviral drugs causes hematologic toxicity particularly in human immunodeficiency viruses (HIV) infection produced³. The azole antifungal agents in clinical use contain either two or three nitrogens in the azole ring and are thereby classified as imidazoles (e.g., ketoconazole, miconazole, clotrimazole) or triazoles (e.g., voriconazole, genaconazole, terconazole, itraconazole, fluconazole) respectively. With the exception of ketoconazole, use of the imidazoles is limited to the treatment of superficial mycoses, whereas the triazoles have a broad range of applications in the treatment of both superficial and systemic fungal infections. Diseases which very recently seemed on their way of extinction, such as fungal and tuberculosis are once again becoming serious public health problems because of societal changes and resistance emergence by pathogens⁴. These developments and the associated increase in microbial infections intensified the search for new, safer and more efficacious agents to combat serious microbial infections. Therefore there is a need to develop novel molecules as antimicrobial, antitubercular and antifungal agents.

The efficiency of triazole derivatives as chemotherapeutic agent is well established and their chemistry has been comprehensively studied. Literature survey revealed that various compounds of 1,2,4-triazoles bearing aryl groups or heterocyclic residues⁵ and condensed heterocyclic derivatives of 1,2,4-triazolo[3,4-b]-1,3,4-thiadiazole rings which possess excellent biological activities, viz. Antibacterial^{6,7}, antifungal⁸, antitubercular^{9,10}, cytotoxic¹¹, anticancer^{12,13}, anti-inflammatory¹⁴⁻¹⁶, analgesic¹⁷, anticonvulsant¹⁸ and anthelmintic¹⁹ activities. In view of these facts and in continuation of our research program on synthesis and biological importance of various heterocyclic compounds²⁰⁻²³, now we are reporting the synthesis and antimicrobial activity of 3,6-disubstituted-1,2,4-triazolo[3,4-b]-1,3,4-thiadiazole derivatives as shown in scheme 1.

Experimental section

All research chemicals were purchased from Acros organics, Sigma-Aldrich, Lancaster Co. and used as such for the reactions. Reactions were monitored by thin layer chromatography (TLC) on pre-coated silica gel plates from E. Merck and Co. Melting points of synthesized compounds were determined in ThermoNIK melting point apparatus and uncorrected and IR spectra were recorded on Thermo Nicolet IR200 FT-IR spectrometer. The ¹H NMR and ¹³C NMR were recorded on Bruker AVANCE II 400. Chemical shifts are reported in δ ppm units with respect to TMS. The mass spectra were recorded using GCMS-QP 2010. A novel series of substituted triazolothiadiazoles 2a-j were synthesized as shown in scheme 1 by the reaction between thiocarbohydrazide and substituted phenoxy acetic acid, which on fusion to form 3-(substituted phenoxy methyl)-4-amino-5-mercapto-1,2,4-triazole (1). General method for synthesis of 3-(substituted phenoxy methyl)-6-phenyl/substituted phenoxy methyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2a-j) Equimolar proportions of N'-4-amino-5-(substituted phenoxy methyl)-2H-1,2,4-triazole-3(4H)-thione (1) and aromatic carboxylic acid or substituted phenoxy acetic acid was

added to 10 ml of dry phosphorous oxy chloride and the solution was refluxed on water bath for 6 hr. Excess of phosphorous oxychloride was removed under vacuum. The thick mass obtained was treated with water and left overnight. Solid thus obtained was filtered washed with 2% sodium carbonate solution, then with cold water, dried and recrystallized from ethanol.

3-phenoxyethyl-6-phenyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2a)

IR (KBr, cm^{-1}): 3048.1 (Ar C-H), 2914.2 (OCH_2), 2747.1 (CH_2), 1587.8 ($\text{C}=\text{N}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.7 (m, 10H, Ar-H), 5.1 (s, 2H, OCH_2).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 175.0 (thiadiazole- C_2), 162.8 (triazole- C_3), 147.9 (triazolothiadiazole- C_5), Ar C [160.2 (C_1), 129.3 (C_3 and C_5), 115.1 (C_2), 114.0 (C_4 and C_6)], Ar C (thiadiazole) [134.6 (C_1), 129.3 (C_3 and C_5), 128.2 (C_4), 127.6 (C_2 and C_6)], 65.8 (OCH_2).

MS (ESI) ($\text{M}+1$) : m/z 309.2; calcd.308.3.

3,6-bis(phenoxyethyl)-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2b)

IR (KBr, cm^{-1}): 3050.1 (Ar C-H), 2910.3 (OCH_2), 2745.5 (CH_2), 1587.8 ($\text{C}=\text{N}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.7 (m, 10H, Ar-H), 5.2 (s, 4H, (OCH_2)₂).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 172.4 (thiadiazole- C_2), 161.9 (triazole- C_3), 147.2 (triazolothiadiazole- C_5), Ar C [153.2 (C_1), 129.9 (C_3 and C_4), 125.2 (C_5), 117.0 (C_2 and C_6)], Ar C (thiadiazole) [158.6 (C_1), 129.3 (C_3 and C_5), 122.1 (C_4), 112.2 (C_2 and C_6)], 65.5 (OCH_2)₂.

MS (ESI) : m/z 338.2; calcd.338.2.

6-[(2,4-dichlorophenoxy)methyl]-3-phenoxyethyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2c)

IR (KBr, cm^{-1}): 3056.2 (Ar C-H), 2911.6 (OCH_2), 2744.3 (CH_2), 1617.5 ($\text{C}=\text{N}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.1-6.6 (m, 8H, Ar-H), 5.1 (s, 4H, (OCH_2)₂).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 163.2 (thiadiazole- C_2), 162.2 (triazole- C_3), 147.3 (triazolothiadiazole- C_5), Ar C [152.4 (C_1), 130.7 (C_3), 129.3 (C_2 and C_4), 128.3 (C_5), 116.8 (C_6)], Ar C (thiadiazole) [156.5 (C_1), 130.4 (C_3 and C_5), 122.6 (C_4), 113.8 (C_2 and C_6)], 66.2 (OCH_2)₂.

MS (ESI) ($\text{M}+1$) : m/z 408.3; calcd.407.2.

3-[(4-chlorophenoxy)methyl]-6-phenyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2d)

IR (KBr, cm^{-1}): 3050.2 (Ar C-H), 2915.6 (OCH_2), 2745.6 (CH_2), 1650.2 ($\text{C}=\text{N}$), 765.2 ($\text{C}-\text{Cl}$).

^1H NMR (DMSO-d_6 , δ ppm): 7.4-6.7 (m, 9H, Ar-H), 5.2 (s, 2H, OCH_2).

MS (ESI) ($\text{M}+1$): m/z 342.7; calcd: 343.8.

3-[(4-chlorophenoxy)methyl]-6-phenoxyethyl-1,2,4-triazolo [3,4-b][1,3,4]thiadiazole (2e)

IR (KBr, cm^{-1}): 3045.1 (Ar C-H), 2912.6 (OCH_2), 2747.5 (CH_2), 1665.8 ($\text{C}=\text{N}$), 766.7 ($\text{C}-\text{Cl}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.7(m, 9H, Ar-H), 5.2 (s, 4H, (OCH_2)₂).

MS (ESI) ($\text{M}+1$) : m/z 373.2; calcd.372.2.

6-[(2,4-dichlorophenoxy)methyl]-3-[(4-chlorophenoxy)methyl]-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2f)

IR (KBr, cm^{-1}): 3050.5 (Ar C-H), 2918.8 (OCH_2), 2740.2 (CH_2), 1660.7 ($\text{C}=\text{N}$), 775.2 ($\text{C}-\text{Cl}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.1-6.6 (m, 7H, Ar-H), 5.1 (s, 4H, (OCH_2)₂).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 163.2 (thiadiazole- C_2), 162.7 (triazole- C_3), 148.0 (triazolothiadiazole- C_5), Ar C [152.7 (C_1), 131.1 (C_3), 130.3 (C_2 and C_4), 128.1 (C_5), 116.4 (C_6)], Ar C (thiadiazole) [153.7 (C_1), 130.4 (C_3), 123.7 (C_4), 127.9 (C_2), 127.7 (C_5), 116.8 (C_6)], 66.7 (OCH_2)₂.

MS (ESI) ($\text{M}+2$) : m/z 443.7; calcd.441.7.

3-[(4-nitrophenoxy)methyl]-6-phenyl-1,2,4-triazolo[3,4-b] [1,3,4] thiadiazole (2g)

IR (KBr, cm^{-1}): 3068.8 (Ar C-H), 2922.1 (OCH_2), 2746.3 (CH_2), 1617.5 ($\text{C}=\text{N}$).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 8.1-7.1 (m, 9H, Ar-H), 5.2 (s, 2H, OCH_2).

MS (ESI) : m/z 453.3; calcd.453.3.

3-[(4-nitrophenoxy)methyl]-6-phenoxyethyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2h)

IR (KBr, cm^{-1}): 3052.1 (Ar C-H), 2915.3 (OCH_2), 1583.3 (C=N).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.7 (m, 8H, Ar-H), 5.2 (s, 4H, (OCH_2)₂).

MS (ESI): m/z 482.9; calcd.483.2.

3-[(2,4-dichlorophenoxy)methyl]-6-phenyl-1,2,4-triazolo[3,4-b][1,3,4]thiadiazole (2i)

IR (KBr, cm^{-1}): 3070.2 (Ar C-H), 2917.3 (OCH_2), 2747.1 (CH_2), 1617.2(C=N).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.6 (m, 10H, Ar-H), 5.2 (s, 2H, OCH_2).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 173.5 (thiadiazole- C_2), 161.2 (triazole- C_3), 147.3 (triazolothiadiazole- C_5), Ar C [153.6 (C_1), 131.1 (C_3), 129.3 (C_2 and C_4), 129.1 (C_5), 116.7 (C_6)], Ar C (thiadiazole) [133.2 (C_1), 130.1 (C_3 and C_5), 128.6 (C_4), 127.2 (C_2 and C_6)], 66.2 (OCH_2).

MS (ESI) ($\text{M}+2$) : m/z 379.2; calcd.377.2.

3-[(2,4-dichlorophenoxy)methyl]-6-phenoxyethyl-1,2,4-triazolo [3,4-b][1,3,4] thiadiazole (2j)

IR (KBr, cm^{-1}): 3050.1 (Ar C-H), 2910.3 (OCH_2), 1587.8 (C=N).

^1H NMR (400 MHz, DMSO-d_6 , δ ppm): 7.4-6.7 (m, 8H, Ar-H), 5.2 (s, 4H, (OCH_2)₂).

^{13}C -NMR (400 MHz, DMSO-d_6 , δ ppm): 163.2 (thiadiazole- C_2), 162.2 (triazole- C_3), 147.3 (triazolothiadiazole- C_5), Ar C [152.4 (C_1), 130.7 (C_3), 129.3 (C_2 and C_4), 128.3 (C_5), 116.8 (C_6)], Ar C (thiadiazole) [156.5 (C_1), 130.4 (C_3 and C_5), 122.6 (C_4), 113.8 (C_2 and C_6)], 66.2 (OCH_2)₂.

In-vitro antimicrobial activity

Antibacterial and antifungal activity of the synthesized compounds were performed by agar dilution method²⁴ using standard bacterial and fungal strains, such as; *Streptococcus aureus* (ATCC 9144), *Bacillus subtilis* (ATCC 6633), *Pseudomonas aeruginosa* (ATCC 25668), *Escherichia coli* (ATCC 25922), *Candida albicans* (ATCC 2091), *Aspergillus niger* (ATCC 6275) and *Aspergillus fumigatus* (ATCC 13073) respectively. Antitubercular activities of the synthesized compounds were performed by broth dilution assay²⁵ by incubated the test compounds at 37°C for 21 days using middle brook 7H9 broth media against *M. tuberculosis* H₃₇R_v strain.

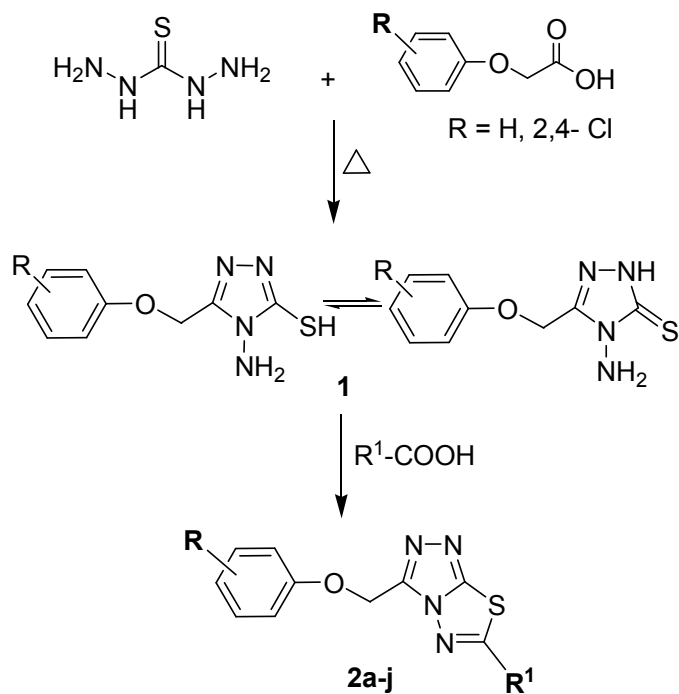
RESULTS AND DISCUSSION

The IR spectrum of compound **1** showed characteristic absorption bands at 3457.4 cm^{-1} was assigned to NH_2 and the other at 1282.1/2580 cm^{-1} was attributed to C=S/SH. These were disappeared by the formation of triazolothiadiazoles 2a-j. Similarly the ^1H NMR characteristic signal singlet showed at δ 13.2/7.8 due to SH/NH proton derived from tautomeric equilibrium of compound **1** and δ 2.1 due to NH_2 protons. Which were absent in the ^1H NMR spectra of triazolothiadiazoles. It suggest that Cyclocondensation of the SH and NH_2 functions of **1** with aromatic carboxylic acid and substituted phenoxy acetic acids resulted triazolothiadiazoles ring. The structure of compounds was further conformed by evidence for the formation of triazolothiadiazoles ring which was obtained by recording mass spectral data. The characterization data of all compounds is tabulated in Table 1.

In-vitro antimicrobial activity

Antibacterial and antifungal screening data revealed that all the compounds showed moderate to good antibacterial and antifungal activity respectively. Antibacterial and antifungal results are given in Table 2. Compounds showed antibacterial and antifungal activity at MIC values of 1-64 $\mu\text{g}/\text{mL}$ respectively. Compounds 2e (R=4-Cl, R¹=C₆H₅OCH₂), 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂) inductively electron withdrawing substituents on phenoxy methyl rings were exhibited equipotent

antibacterial activity against *S. aureus*, *B. subtilis* at MIC of 1 µg/mL whereas these compounds showed comparable activity against gram negative bacteria *P. aeruginosa*, *E.coli* at MIC of 2 µg/mL. Compounds 2c (R=H, R¹=2,4-ClC₆H₃OCH₂), 2d (R=4-Cl, R¹=C₆H₅) showed comparable antibacterial activity against gram positive bacteria *S. aureus*, *B. subtilis* at MIC of 2 µg/mL. Compounds 2j (R=4-Cl, R¹= C₆H₅OCH₂), 2d (R=4-Cl, R¹=C₆H₅) showed comparable antibacterial activity against *E.coli* at MIC of 2 µg/mL. Rest of the compounds showed weak activity against both gram positive (*S. aureus*, *B. subtilis*) and gram negative (*P. aeruginosa*, *E. coli*) bacteria, when compared with standard drug Ceftriaxone which showed MIC of 1µg/mL. The compound 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂) having electron withdrawing moiety on phenoxy methyl rings were found to be equipotent antifungal activity against *A. niger* and *A. fumigatus* at MIC of 1 µg/mL. Compounds 2d (R=4-Cl, R¹=C₆H₅), 2i (R=2,4-Cl, R¹=C₆H₅) showed equipotent antifungal activity against *A. niger* at MIC of 1 µg/mL. Whereas compounds 2b (R=H, R¹=C₆H₅OCH₂), 2c (R=H, R¹=2,4-ClC₆H₃OCH₂), 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂) were exhibited equipotent antifungal activity against *A. fumigatus* and compounds 2b (R=H, R¹=C₆H₅OCH₂), 2c (R=H, R¹=2,4-ClC₆H₃OCH₂), 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂) were showed equipotent antifungal activity against *A. fumigatus* at MIC of 1 µg/mL respectively. The compounds 2e (R=4-Cl, R¹=C₆H₅OCH₂), 2j (R=4-Cl, R¹= C₆H₅OCH₂) showed equipotent antifungal activity against *A. niger* and *A. fumigatus* at MIC of 2 µg/mL. Out of ten synthesized compounds, compound 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂) showed comparable antifungal activity against *C. albican*. In contrary, all compounds have shown less antifungal activity against *C. Albicans* when compared with standard drug Fluconazole which showed at MIC of 1 µg/mL. In general these compounds are found to possess more antifungal activity than antibacterial activity. Hence, it reveals that the compounds having electron withdrawing substituents on phenoxy methyl rings were found to be more active. The compounds showed antitubercular activity at MIC values of 0.25-2 µg/mL. From the results the compounds 2c (R=H, R¹=2,4-ClC₆H₃OCH₂), 2f (R=4-Cl, R¹=2,4-ClC₆H₃OCH₂), 2j (R=4-Cl, R¹= C₆H₅OCH₂) having chloro groups on phoxymethyl rings were showed comparable antitubercular activity against *M. tuberculosis* H₃₇R_v at MIC of 0.50 µg/mL, when compared with standard drug Rifampin and INH which showed MIC of 0.25 µg/mL. Rest of the compounds in the series showed moderate antitubercular activity. The results are shown in Table 2.



Scheme 1

Table 1
Characterization data of 2a-j

Compound	R	R ¹	Yield (%)	Melting point* (°C)	R _f Value [#]	Mol. formula
2a	H	C ₆ H ₆	60.0	153-154	0.54	C ₁₆ H ₁₂ N ₄ O ₂ S
2b	H	C ₆ H ₅ OCH ₂	62.5	212-214	0.33	C ₁₇ H ₁₄ N ₄ O ₂ S
2c	H	2,4-ClC ₆ H ₃ OCH ₂	54.0	235-237	0.40	C ₁₇ H ₁₂ Cl ₂ N ₄ O ₂ S
2d	4-Cl	C ₆ H ₆	60.0	198-200	0.52	C ₁₆ H ₁₁ ClN ₄ O ₂ S
2e	4-Cl	C ₆ H ₅ OCH ₂	59.0	188-190	0.53	C ₁₇ H ₁₁ ClN ₄ O ₂ S
2f	4-Cl	2,4-ClC ₆ H ₃ OCH ₂	58.0	203-206	0.49	C ₁₇ H ₁₁ ClN ₄ O ₂ S
2g	4-NO ₂	C ₆ H ₆	63.2	186-188	0.39	C ₁₆ H ₁₁ N ₅ O ₃ S
2h	4-NO ₂	C ₆ H ₅ OCH ₂	60.2	190-192	0.42	C ₁₇ H ₁₃ N ₅ O ₄ S
2i	2,4-Cl	C ₆ H ₆	62.3	189-191	0.35	C ₁₆ H ₁₀ Cl ₂ N ₄ O ₂ S
2j	2,4-Cl	C ₆ H ₅ OCH ₂	63.8	218-220	0.47	C ₁₇ H ₁₂ Cl ₂ N ₄ O ₂ S

The C, H, N analysis was found to be ± 0.4 %.

** Recrystallization solvent ethanol.*

#Stationary Phase: Silica gel G, Mobile phase: Chloroform: Benzene (1:1), Iodine vapors as visualizing agent.

Table 2
***In-vitro* antimicrobial activity of 2a-j**

Compound	R	R ¹	MIC values (µg/mL)							
			Sa ^a	Bs ^a	Pa ^b	Ec ^b	Ca ^c	An ^c	Af ^c	Mt ^d
2a	H	C ₆ H ₅	16	8	8	32	32	64	16	1
2b	H	C ₆ H ₅ OCH ₂	16	16	64	64	32	64	1	1
2c	H	2,4-ClC ₆ H ₃ OCH ₂	2	2	8	8	64	64	1	0.50
2d	4-Cl	C ₆ H ₅	2	2	8	16	64	1	16	1
2e	4-Cl	C ₆ H ₅ OCH ₂	1	1	2	4	32	2	2	2
2f	4-Cl	2,4-ClC ₆ H ₃ OCH ₂	1	1	2	2	2	1	1	0.50
2g	4-NO ₂	C ₆ H ₅	16	16	8	8	32	32	64	2
2h	4-NO ₂	C ₆ H ₅ OCH ₂	8	16	8	16	64	16	32	2
2i	2,4-Cl	C ₆ H ₅	4	4	8	16	64	1	64	1
2j	2,4-Cl	C ₆ H ₅ OCH ₂	4	8	4	2	32	2	2	0.50
Ceftriaxone			1	1	1	1	ND	ND	ND	ND
Fluconazole			ND	ND	ND	ND	1	1	1	ND
Rifampin			ND	ND	ND	ND	ND	ND	ND	0.25
INH			ND	ND	ND	ND	ND	ND	ND	0.25

^a The screening organism. Gram-positive bacteria: *Streptococcus aureus* (ATCC 9144, Sa), *Bacillus subtilis* (ATCC 6633, Bs).

^b The screening organisms. Gram-negative bacteria: *Pseudomonas aeruginosa* (ATCC 25668, Pa), *Escherichia coli* (ATCC 2091, Ec).

^c The screening fungal organisms: *Candida albicans* (ATCC 2091, Ca), *Aspergillus niger* (ATCC 6275, An), *Aspergillus fumigatus* (ATCC 13073, Af).

^d The screening organisms: *Mycobacterium tuberculosis* H₃₇R_v (Mt); ND: Not determined.

CONCLUSION

In conclusion the synthesized title compounds resulted in good yields. The SAR study of the title compounds reveals better activity. Indicating the triazolothiadiazole ring scaffold influences the pharmacological activity. From the activity data, the compounds having electron withdrawing group such as halogens are found to be more activity as compared to others. It is observed that the chloro at *ortho* and *para* position on phoxymethyl rings increases the activity. However, the difference in activity profile with structural modifications provides further scope to explore these compounds for better bioactivity. Thus substituted triazolothiadiazole moieties appear to be another interesting source of antifungal and anti-tubercular agents. Further studies are required to

know the mechanism of action of these compounds. The authors have declared no conflict of interest.

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